



# Pinnacle

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## P-SAT 2025

Pinnacle Scholastic Aptitude Test

For Classes 6<sup>th</sup> to 11<sup>th</sup> (Appearing)

### Pinnacle Scholastic Aptitude Test

# SAMPLE TEST PAPER

## Class 11<sup>th</sup> NM

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# INSTRUCTIONS

**Duration: 2 Hour**

**Maximum Marks: 240**

Please read the instructions carefully. You are allotted 5 minutes specifically for this purpose.

Things NOT ALLOWED in EXAM HALL: Blank Paper, clipboard, log table, slide rule, calculator, camera, mobile and any electronic or electrical gadget. If you are carrying any of these, then keep them at a place specified by invigilator at your own responsibility.

## I. General Instructions

1. This Booklet is your question paper. Don't break the seal of booklet until the invigilator instructs to do so.
2. The answer sheet (ORS) is provided to you separately which is a machine readable optical response sheet. You have to mark your answers in the ORS by darkening bubbles as per your answer choice, by using black or blue ball point pen.
3. This question paper contains three sections as follows:
  - a) **Section - A** contains total 20 questions (Q. No. 1 to 20) of Physics.
  - b) **Section - B** contains total 20 questions (Q. No. 21 to 40) of Chemistry.
  - c) **Section - C** contains total 20 questions (Q. No. 41 to 60) of Mathematics.
4. Rough spaces are provided for rough work inside the question paper. No additional sheets will be provided for rough work.
5. If you are found involved in cheating or disturbing others, then your ORS will be cancelled.

## II. Filling of OMR Sheet

1. Ensure matching of OMR sheet with the Question paper before you start marking your answers on OMR sheet.
2. On OMR sheet, darken the appropriate bubble with black/blue pen for each character of your Enrollment No. and your Name, Test Code and other details at the designated places.
3. Think wisely before darkening bubble as there is negative marking for wrong answer. Answer once marked by pen can't be cancelled.
4. Don't put any stain on ORS and hand it over back properly to the invigilator.

## III. Marking Scheme

1. All the questions of **each Part** have only one correct answer.
2. If darken bubble is **RIGHT** answer: 4 Marks.
3. If darken bubble is **WRONG** answer: -1 Marks (Minus One Mark).
4. If no bubble is darkened in any question: No Marks.

Please follow the instructions





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### NEET RESULT



|   |  |  |                                   |
|---|--|--|-----------------------------------|
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# What Makes Pinnacle Unique?

- 1 Best Faculty Team
- 2 Hi-Tech Classrooms
- 3 Customized Study material
- 4 Personal Attention
- 5 Unlimited Doubt Sessions
- 6 Best Testing Methodology



Lecture Recording Lab



Revision Classes & Self Study Zones



Online Testing Lab (CBT)



Smart & AC Classrooms



Objective + Subjective



Student Performance and Analysis Report



Rich Library & Researched Study Modules



Extra Support to weaker students



CCTV Campus



Biometric Attendance

## Classroom Program

### 1 Comprehensive Classroom Lectures

All classes at Pinnacle are conducted by highly qualified and experienced faculty members, mostly IITians. Each chapter is started at the grass root level and is dealt to an extent which is the requirement of competitive examinations, with an aim of enabling the students to develop a comprehensive view of the whole chapter with a thorough understanding.



### Doubt Clearance 2

"If you ask a question, you may appear fool for some time, but if you don't, you'll remain a fool for whole life." System at Pinnacle encourages all students to ask their doubts and questions.

### 3 Regular Tests Online and Offline

As JEE Mains and Advanced have gone completely online and NEET is in the pipeline, we have launched a dedicated online testing platform where students can practise over CBT (Computer Based Tests). The combination of online and offline testing modes based on latest JEE/NEET patterns ensure that students are at par with the recent changes. Students can check their test reports and performance analysis via a unique online login ID. Their results are also communicated to parents via SMS.

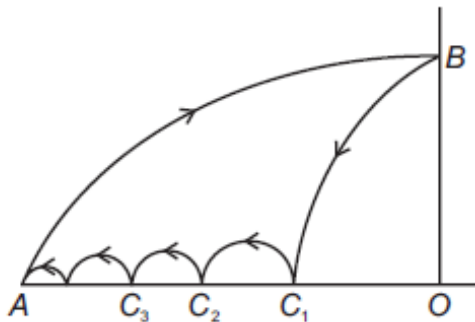


### Addressing the Board Exam 4

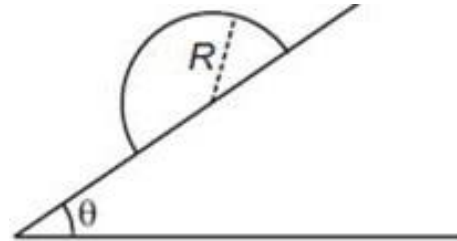
Pinnacle has a very distinct methodology for preparing the students for competitive examinations while in full synchronization with Board Exams as well. Board level tests are conducted alongside the regular JEE/NEET tests and the copies are graded at very meticulous level by teachers. Students receive methodological tips so as to perform excellent in the board Exams as well.

**Section - A**  
**Physics**

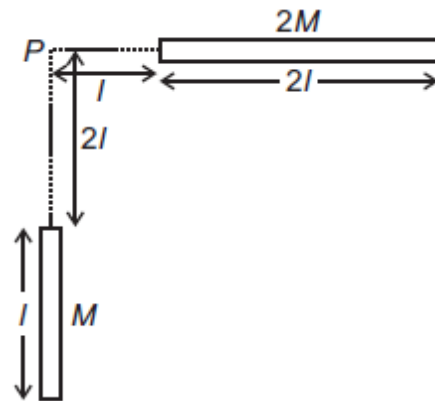
1. A ball was projected from point  $A$  and it hit a vertical wall at point  $B$  horizontally. From  $B$  the ball flies to point  $C_1, C_2, \dots$  and finally comes back to the point  $A$ . The coefficient of restitution is  $e$  for every collision. Find  $e$ , if the ball does not bounce up at all when it gets back to the point  $A$



- (a)  $\frac{\sqrt{2}-1}{2}$   
 (b)  $2-\sqrt{3}$   
 (c)  $\sqrt{2}-1$   
 (d)  $\frac{2}{\sqrt{2}+1}$
2. A uniform hemisphere of radius  $R$  is placed on rough inclined plane. The coefficient of the friction between the surfaces is  $\mu$ . The angle of inclination ( $\theta$ ) of the plane is increased. The block will



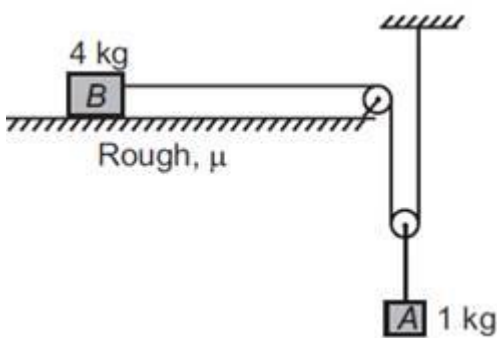
- (a) Topple before sliding if  $\mu \geq \frac{8}{3}$   
 (b) Topple before sliding if  $\mu \geq \frac{4}{3}$   
 (c) Slide before toppling if  $\mu \geq \frac{8}{3}$   
 (d) Slide before toppling if  $\mu \geq 3$
3. A rod of length  $l$  and of mass  $M$  and another rod of length  $2l$  and mass  $2M$  are arranged as shown below in the figure. The magnitude of gravitational field at point  $P$  is



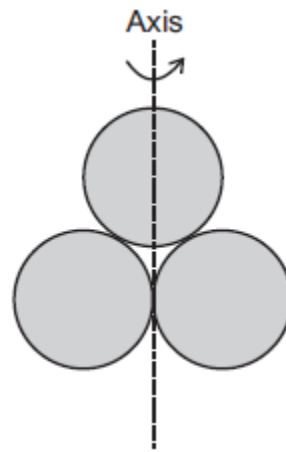
*Space for Rough Work*

- (a)  $\frac{\sqrt{5} GM}{6 l^2}$
- (b)  $\frac{\sqrt{65} GM}{6 l^2}$
- (c)  $\frac{\sqrt{17} GM}{6 l^2}$
- (d)  $\frac{3GM}{\sqrt{18}l^2}$

4. In the given arrangement, the pulleys and strings are ideal. The blocks are released from rest and it is found that block A has speed 0.3 m/s after it has descended through a distance of 1 m. The coefficient of friction ( $\mu$ ) is ( $g = 10 \text{ m/s}^2$ )



- (a) 0.28
  - (b) 0.12
  - (c) 0.32
  - (d) 0.21
5. Three discs each of mass  $m$  and radius  $R$  are joined in a plane as shown in figure. The radius of gyration of this system about an axis indicated in the figure is

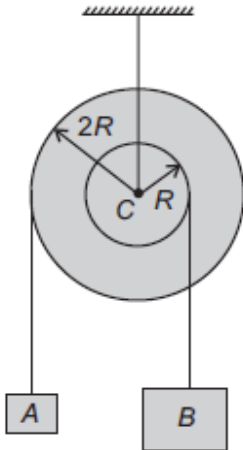


- (a)  $\sqrt{\frac{7}{12}}R$
  - (b)  $\sqrt{\frac{11}{12}}R$
  - (c)  $\sqrt{\frac{7}{16}}R$
  - (d)  $\sqrt{\frac{5}{6}}R$
6. Three identical particles each of mass  $m$  are at the vertices of an equilateral triangle of side length  $l$ . Two particles are fixed and third particle is released from rest under their gravitational force alone. The maximum speed of third particle in subsequent motion is given by
- (a)  $2\sqrt{\frac{Gm}{l}}$
  - (b)  $\sqrt{\frac{Gm}{l}}$
  - (c)  $\sqrt{\frac{2Gm}{l}}$

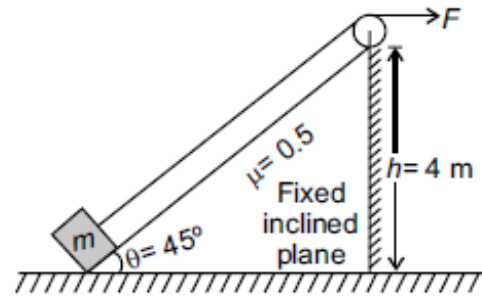
Space for Rough Work

(d)  $\sqrt{\frac{Gm}{2l}}$

7. In the given arrangement, a spool of moment of inertia  $I_c = 4 mR^2$  is connected with two blocks A and B of masses  $2m$  and  $3m$  respectively and released. Assuming there is no slipping anywhere, then the acceleration of block A is



- (a)  $\frac{g}{5}$   
 (b)  $\frac{2g}{7}$   
 (c)  $\frac{2g}{15}$   
 (d)  $\frac{4g}{13}$
8. A block of mass  $m = 4 \text{ kg}$  is placed near the bottom of a rough ( $\mu = 0.5$ ) inclined plane as shown in figure. The minimum work done by applied force required to raise the block slowly to the top of inclined is ( $g = 10 \text{ m/s}^2$ )



- (a) 80 J  
 (b) 160 J  
 (c) 240 J  
 (d) 320 J
9. A uniform rod of mass  $M$ , length  $L$  and cross-sectional area  $A$  moves under the action of two forces on a smooth surface as shown in figure. If Young's modulus of material of rod is  $Y$ , then elastic potential energy stored in the rod is
- 
- (a)  $\frac{7FL^2}{3YA}$   
 (b)  $\frac{3F^2L}{8YA}$   
 (c)  $\frac{7F^2L}{6YA}$   
 (d)  $\frac{13F^2L}{7YA}$
10. An ideal gas whose adiabatic exponent equals  $\gamma$  is expanded so that  $\Delta Q + \Delta U = 0$ . Then the equation of the process in the variables  $T$  and  $V$  is
- (a)  $TV^{\frac{(\gamma-1)}{2}} = C$

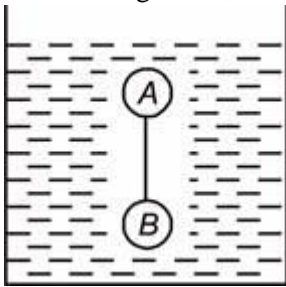
Space for Rough Work

(b)  $TV \frac{(\gamma - 2)}{2} = C$

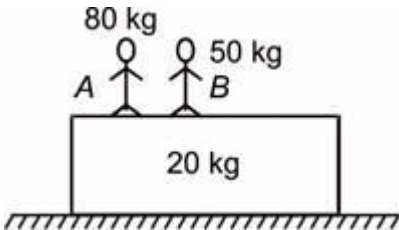
(c)  $TV \frac{(\gamma - 1)}{4} = C$

(d)  $TV \frac{(\gamma - 2)}{4} = C$

11. Two solid spheres *A* and *B* of equal volume but different densities, are connected with a string and placed inside a liquid. They are at equilibrium in the position shown in the figure. If the string is cut, then



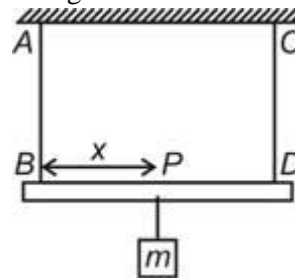
- (a) *A* may remain at rest  
 (b) *B* may remain at rest  
 (c) *A* may move upwards with acceleration more than *g*  
 (d) *B* may move downwards with acceleration more than *g*
12. Two persons, *A* of mass 80 kg and *B* of mass 50 kg are standing on a horizontal platform of mass 20 kg. The platform is on horizontal frictionless surface and is initially at rest



If both *A* and *B* jump (horizontally) from platform simultaneously and in same direction

with 2 m/s each, what would be the velocity of centre of mass of the system of persons and platform?

- (a) 1.4 m/s  
 (b) 3.2 m/s  
 (c) 2.8 m/s  
 (d) Zero
13. A massless rod *BD* (length = *l*) is suspended by two identical massless strings *AB* and *CD* of equal lengths. A block of mass *m* is suspended at point *P*. If the fundamental frequency of the left wire is 3 times the fundamental frequency of right wire then determine the value of *x*



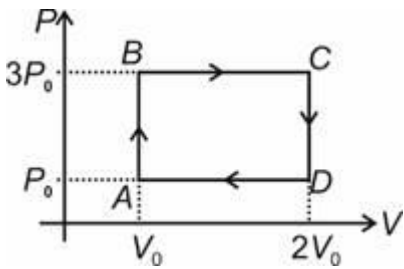
- (a)  $\frac{l}{5}$   
 (b)  $\frac{3l}{5}$   
 (c)  $\frac{9l}{10}$   
 (d)  $\frac{l}{10}$
14. One mole of an ideal gas  $\left(\frac{C_p}{C_v} = \frac{4}{3}\right)$  undergoes a cyclic process *ABCD*, shown by the following indicator diagram. The pressure and volume at certain points have been marked. The

**Space for Rough Work**



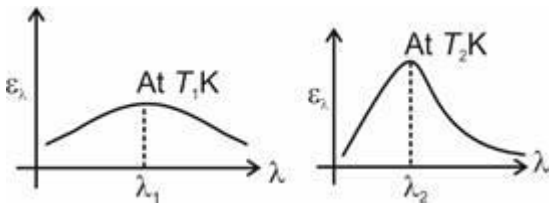
temperature of the gas at state A is  $T_0$ , given by

$$\text{the relation } T_0 = \frac{P_0 V_0}{R}$$



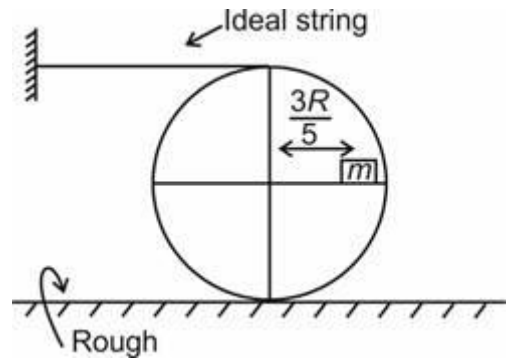
The work done by the gas in one complete cycle is

- (a)  $P_0 V_0$
  - (b)  $2P_0 V_0$
  - (c)  $3P_0 V_0$
  - (d)  $4P_0 V_0$
15. Graph between spectral emissive power ( $\epsilon_\lambda$ ) and wavelength  $\lambda$  for a black body at temperatures  $T_1$ K and  $T_2$ K as shown in figure



If area of  $\epsilon_\lambda - \lambda$  graph at  $T_2$ K is 4 times the area at  $T_1$ K, then

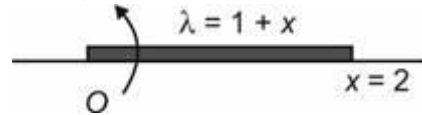
- (a)  $T_1 : T_2 = 1 : \sqrt{2}$
  - (b)  $T_1 : T_2 = 1 : 2$
  - (c)  $\lambda_1 : \lambda_2 = 2 : 1$
  - (d)  $\lambda_1 : \lambda_2 = 1 : \sqrt{2}$
16. Consider a ring of mass  $m$  and radius  $R$  with four light spokes. A point mass  $m$  is placed on one horizontal spoke. (see the figure)



If the shown system is at rest then minimum value of coefficient of friction will be

- (a)  $\frac{3}{20}$
- (b)  $\frac{1}{10}$
- (c)  $\frac{1}{4}$
- (d)  $\frac{2}{5}$

17. A rod of length 2 m has uniform cross-section but linear mass density varies as  $\lambda = (1 + x)$  kg/m. It is placed as shown in figure. The moment of inertia of rod about an axis passing through  $O$  and perpendicular to rod is ( $x$  is in metre)



- (a)  $\frac{5}{3} \text{ kg m}^2$
- (b)  $\frac{20}{3} \text{ kg m}^2$
- (c)  $\frac{3}{5} \text{ kg m}^2$

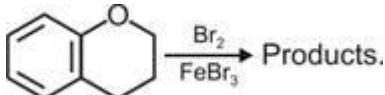
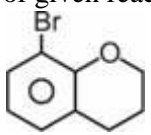
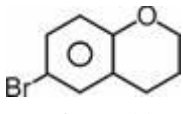
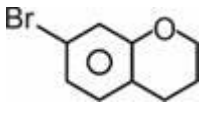
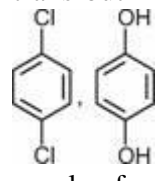
Space for Rough Work

- (d)  $\frac{8}{3} \text{ kg m}^2$
18. A trolley accelerates uniformly from rest and acquires the speed of  $72 \text{ km h}^{-1}$  on moving 200 m distance, henceforth breaks are applied so that it retards the trolley at a rate which is proportional to the velocity, until its speed is halved, by covering 400 m distance further. The total time of motion described of the trolley from the start (approximately)
- (a) 60 s  
(b) 68 s  
(c) 48 s  
(d) 42 s
19. An ideal gas undergoes a thermodynamic process in which internal energy of the gas depends on pressure of the gas as  $U = aP^4$ , where  $a$  is a positive constant. Assuming gas to be monoatomic, the molar heat capacity of the gas for given process will be
- (a)  $\frac{3R}{4}$   
(b)  $\frac{2R}{3}$   
(c)  $\frac{9R}{4}$   
(d)  $\frac{4R}{9}$
20. For an SHM oscillator, the amplitude is 5 cm and its time period is 4 seconds. The minimum time taken by the particle to pass between points which are at distances 4 cm and 3 cm from the centre of oscillation on the same side of it will be
- (a) 0.13 second  
(b) 0.18 second  
(c) 0.26 second  
(d) 0.35 second

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*Space for Rough Work*

## Section - B Chemistry

21. 250 mL of  $\text{KHC}_2\text{O}_4$  required 100 mL of 0.1 M  $\text{KMnO}_4$  solution in acidic medium for complete oxidation. If the same 250 mL of  $\text{KHC}_2\text{O}_4$  is neutralised completely using 0.2 M KOH solution, then the volume of KOH solution (in mL) used is
- 100 mL
  - 50 mL
  - 125 mL
  - 150 mL
22.  Products of given reaction is/are
- Only 
  - Only 
  - Mixture of a and b
  - Only 
23. At  $20^\circ\text{C}$ , the concentration of  $\text{Ag}^+$  ions in a saturated solution of  $\text{Ag}_2\text{CrO}_4$  is  $1.5 \times 10^{-4}$  mol/L. The solubility product of  $\text{Ag}_2\text{CrO}_4$  at  $20^\circ\text{C}$  will be
- $2.25 \times 10^{-8}$
  - $1.687 \times 10^{-12}$
  - $1.658 \times 10^{-13}$
  - $13.5 \times 10^{-16}$
24. The correct order of increasing  $s$ -character (in percentage) in the hybrid orbitals of central atom in following molecules/ions is
- $\text{CO}_3^{2-}$
  - $\text{XeF}_4$
  - $\text{I}_3^-$
  - $\text{NO}_2^+$
- $\text{II} < \text{III} < \text{IV} < \text{I} < \text{V}$
  - $\text{II} < \text{IV} < \text{III} < \text{V} < \text{I}$
  - $\text{III} < \text{II} < \text{I} < \text{V} < \text{IV}$
  - $\text{II} < \text{IV} < \text{III} < \text{I} < \text{V}$
25. In which pair the dipole moment of first species is greater than second?
- $\text{NH}_3$ ,  $\text{NF}_3$
  - $\text{BF}_3$ ,  $\text{BCl}_3$
  - trans-but-2-ene, cis-but-2-ene
  - 
26. Two mole of an ideal monoatomic gas expands isothermally and reversibly from 10 L to 50 L at  $27^\circ\text{C}$ . From the following select the correct option (Use  $\ln 5 = 1.6$ , notation have their usual meaning)
- $W = -0.96$  kcal
  - $\Delta S_{\text{Total}} = -3.2$  cal/K
  - $\Delta S_{\text{sys}} = 6.4$  cal/K
  - $q_{\text{sys}} = -0.96$  kcal
27. The Prussian blue colour obtained during the test of nitrogen by Lassaigne's test is due to the formation of
- $\text{Fe}_4[\text{Fe}(\text{CN})_6]_3$

*Space for Rough Work*

- (b)  $\text{Na}_3[\text{Fe}(\text{CN})_6]$   
 (c)  $\text{Fe}_3[\text{Fe}(\text{CN})_6]_2$   
 (d)  $\text{Na}_4[\text{Fe}(\text{CN})_5\text{NOS}]$   
 Ans.  $\text{Fe}_4[\text{Fe}(\text{CN})_6]_3$
28. If the radius of first Bohr orbit for hydrogen atom is  $x$ , then de-Broglie wavelength of electron in 4<sup>th</sup> orbit of H atom is nearly  
 (a)  $\pi x$   
 (b)  $2\pi x$   
 (c)  $6\pi x$   
 (d)  $8\pi x$
29.  $\text{A} + 2\text{B} \rightleftharpoons 2\text{C} + \text{D}$   
 The initial conc. of B is 1.5 times the initial conc. of A. At equilibrium, conc. of A is equal to B. The equilibrium constant  $K_c$  is  
 (a) 2  
 (b) 8  
 (c) 4  
 (d) 9
30. Ozone layer is present in  
 (a) Thermosphere  
 (b) Mesosphere  
 (c) Stratosphere  
 (d) Ionosphere
31. Bond order of  $\text{CO}^+$  and  $\text{N}_2^+$  are respectively  
 (a) 3.5 and 2.5  
 (b) 2.5 and 2.5  
 (c) 3 and 2.5  
 (d) 2.5 and 3
32. An open vessel at  $27^\circ\text{C}$  contains 28 grams of  $\text{H}_2$ . Its temperature increases upto  $127^\circ\text{C}$ . What fraction of moles of  $\text{H}_2$  gas escapes out of vessel?  
 (a)  $\frac{1}{8}$   
 (b)  $\frac{1}{4}$   
 (c)  $\frac{2}{3}$   
 (d)  $\frac{3}{4}$
33. A compound of boron X reacts at high temperature with  $\text{NH}_3$  to give another compound Y, which is colourless liquid and is isosteric with benzene. X on reaction with excess  $\text{NH}_3$  gives  $(\text{BN})_n$ . X and Y respectively are  
 (a)  $\text{BH}_3$ ,  $\text{B}_2\text{H}_6$   
 (b)  $\text{NaBH}_4$ ,  $\text{C}_6\text{H}_6$   
 (c)  $\text{B}_2\text{H}_6$ ,  $\text{B}_3\text{N}_3\text{H}_6$   
 (d)  $\text{B}_4\text{C}_3$ ,  $\text{B}_3\text{N}_3\text{H}_6$
34. The molarity of 20 ml of 10 V  $\text{H}_2\text{O}_2$  is  
 (a) 0.45 M  
 (b) 0.89 M  
 (c) 1.33 M  
 (d) 11.2 M
35. For the reaction  

$$\text{CH}_3 - \text{CH}_2 - \text{CH} = \text{CH}_2 \xrightarrow{\text{Br}_2/\text{CCl}_4} \text{X} \xrightarrow{2\text{NaNH}_2/\Delta} \text{Y}$$
 (P)  
 The incorrect statement is  
 (a) y has same degree of unsaturation as that of reactant (P)  
 (b) x is formed by anti-addition  
 (c)  $\text{pK}_a$  of y is less than that of reactant (P)  
 (d) x is vicinal di-halide
36. Consider the following reactions  

$$\text{CH}_3 - \text{CH}_2 - \text{CH} = \text{CH}_2 \begin{cases} \xrightarrow{\text{HCl/Peroxide}} \text{X} \\ \xrightarrow{\text{HBr/Peroxide}} \text{Y} \end{cases}$$
  
 If X and Y represents the major product of the respective reaction then select the correct statement.

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- (a) Both X and Y are formed by radical mechanism  
(b) X is formed by radical mechanism while Y is formed by ionic mechanism  
(c) Both X and Y are 2°-alkyl halide  
(d) X is 2°-alkyl halide while Y is 1°-alkyl halide
37. The equivalent mass of  $\text{Na}_2\text{S}_2\text{O}_3$  (molar mass = M) when it reacts with  $\text{I}_2$  in acidic medium is  
(a) M  
(b)  $\frac{M}{8}$   
(c)  $\frac{M}{2}$   
(d)  $\frac{M}{4}$
38.  $\text{S}_8(\text{s}) + 12\text{OH}^-(\text{aq}) \rightarrow 4\text{S}^{2-}(\text{aq}) + 2\text{S}_2\text{O}_3^{2-} + 6\text{H}_2\text{O}(\text{l})$  belongs to which class of redox reactions?  
(a) Combination reaction  
(b) Decomposition reaction  
(c) Disproportionation reaction  
(d) Displacement reaction
39. How many spectral lines are emitted from a sample of hydrogen gaseous atoms when electrons present in  $n^{\text{th}}$  energy level return to ground state?  
(a)  $n(n+2)$   
(b)  $\frac{n(n-1)}{2n}$   
(c)  $\frac{n(n-1)}{2}$   
(d)  $\frac{n(n+2)}{2}$
40.  $\text{CFCl}_3$  is responsible for the decomposition of ozone to oxygen. Which of the following species reacts with ozone to form dioxygen?  
(a)  $\text{Cl}_2$   
(b)  $\text{Cl}^-$   
(c)  $\text{Cl}^\bullet$   
(d)  $\text{F}^-$

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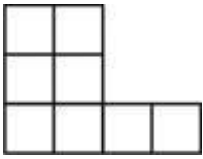
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## Section – C Mathematics

41. If the equations  $ax^2 + 2bx + 3c = 0$  and  $3x^2 + 8x + 15 = 0$  have a common root, where  $a, b, c$  are the lengths of sides of a triangle  $ABC$ , then  $(\sin^2 A + \sin^2 B + \sin^2 C)$  is equal to
- (a) 1  
(b)  $\frac{3}{2}$   
(c)  $\frac{5}{2}$   
(d) 2
42. Triangles are formed by joining the vertices of a regular polygon of 20 sides. The probability that no side of the polygon is a side of the triangle, is
- (a)  $\frac{25}{57}$   
(b)  $\frac{30}{57}$   
(c)  $\frac{35}{57}$   
(d)  $\frac{40}{57}$
43. The value of  $\lim_{n \rightarrow \infty} \sum_{r=1}^n \frac{r}{r^4 + r^2 + 1}$  is equal to
- (a) Zero  
(b)  $\frac{1}{3}$   
(c)  $\frac{1}{2}$   
(d) 1
44. The focus of the parabola  $(y - 3)^2 = 8(x + 2)$  is
- (a) (2, 0)  
(b) (0, 3)  
(c) (0, 2)  
(d) (3, 0)
45. The centres of the circles  $x^2 + y^2 = 1$ ,  $x^2 + y^2 + 6x - 2y - 1 = 0$  and  $x^2 + y^2 - 12x + 4y - 1 = 0$  are
- (a) The vertices of an equilateral triangle  
(b) The vertices of a right angled triangle  
(c) The vertices of an isosceles triangle  
(d) Collinear
46. The normal to the rectangular hyperbola  $xy = 4$  at the point  $t_1$  meets the curve again at the point  $t_2$ . Then the value of  $t_1^3 t_2$  is
- (a) 1  
(b) 2  
(c) -2  
(d) -1
47. The locus of the point representing the complex number  $z$  for which  $|z + 10i|^2 - |z - 10i|^2 = 20$ , is (where  $i = \sqrt{-1}$ )
- (a) A circle  
(b) A straight line  
(c) A parabola  
(d) A hyperbola

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48. The number of ways in which the letters of the word RAMESH can be placed in the squares of the given figure so that no row remains empty, is



- (a) 17280
  - (b) 18720
  - (c) 15840
  - (d) 14400
49. The sum of the series  ${}^{2020}C_0 - {}^{2020}C_1 + {}^{2020}C_2 - {}^{2020}C_3 + \dots + {}^{2020}C_{1010}$  is

- (a)  $\frac{1}{2} {}^{2020}C_{1010}$
- (b)  ${}^{2020}C_{1010}$
- (c) Zero
- (d)  $\frac{-1}{2} {}^{2020}C_{1010}$

50. If

$$\sum_{i=1}^9 (x_i - 5) = 9 \text{ and } \sum_{i=1}^9 (x_i - 5)^2 = 45,$$

then the standard deviation of the nine items  $x_1, x_2, \dots, x_9$  is

- (a) 2
  - (b) 3
  - (c) 4
  - (d) 9
51. The domain

$$\text{of } f(x) = e^{\sin(x-[x])} + [x] \cos\left(\frac{\pi}{[x+1]}\right), \text{ where}$$

$[\cdot]$  represents greatest integer function, is

- (a) R
- (b)  $R - [-1, 0]$

- (c)  $R - [0, 1]$
- (d)  $R - [-1, 0]$

52. If  $1 + \cos \alpha + \cos^2 \alpha + \cos^3 \alpha + \dots + \infty = 2 - \sqrt{2}$ , ( $\pi < \alpha < 2\pi$ ), then the value of  $\alpha$  is

- (a)  $\frac{5\pi}{4}$
- (b)  $\frac{7\pi}{4}$
- (c)  $\frac{4\pi}{3}$
- (d)  $\frac{5\pi}{3}$

53. Which of the following propositions is a contradiction?

- (a)  $(\sim p \vee \sim q) \vee (p \vee \sim q)$
- (b)  $(p \rightarrow q) \vee (p \wedge \sim q)$
- (c)  $(\sim p \wedge q) \wedge (\sim q)$
- (d)  $(\sim p \wedge q) \vee (\sim q)$

54. The point A is symmetric to  $B(4, -1)$  with respect to the bisector of the first quadrant then AB is equal to

- (a) 5 units
- (b)  $4\sqrt{5}$  units
- (c)  $5\sqrt{2}$  units
- (d)  $2\sqrt{5}$  units

55. Let p and q be two statements. Amongst the following, the statement that is equivalent to  $p \rightarrow q$ , is

- (a)  $p \wedge \sim q$
- (b)  $\sim p \wedge q$
- (c)  $\sim p \vee q$
- (d)  $p \vee \sim q$

56. If A and B are the vertices of the triangle ABC and are given by (2, 5) and (4, -

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- 11) respectively, and  $C$  moves along the line  $L \equiv -9x + 7y + 4 = 0$ , then the locus of the centroid of the triangle  $ABC$  is
- (a)  $9x + 7y - 8 = 0$   
 (b)  $27x + 21y - 8 = 0$   
 (c)  $7x - 9y - 8 = 0$   
 (d)  $27x - 21y = 100$
57. The equation of the curve whose parametric equation is  $x = 2t - 3$  and  $y = 4t^2 - 1$ , is given by
- (a)  $y^2 + 6x - y + 8 = 0$   
 (b)  $x^2 + 6x - y + 8 = 0$   
 (c)  $x^2 + y^2 - 6x - y - 8 = 0$   
 (d)  $x^2 + y^2 - 6x + y - 8 = 0$
58. The equation  $\frac{x^2}{16-x} + \frac{y^2}{k-9} = 1$  represents an ellipse if
- (a)  $k > 9$   
 (b)  $k > 16$   
 (c)  $k \in (9, 16) - \left\{ \frac{25}{2} \right\}$   
 (d)  $k \in (-16, -9) - \left\{ -\frac{25}{2} \right\}$
59. Let  $\alpha, \beta$  be real and  $z$  be a non-real complex number. If  $z^2 + \alpha z + \beta = 0$  has two distinct roots on line  $\text{Re}(z) = 1$ , then it is necessary that
- (a)  $\beta \in (-1, 0)$   
 (b)  $|\beta| = 1$   
 (c)  $\beta \in (1, \infty)$   
 (d)  $\beta \in (0, 1)$
60. The value of  $\lim_{x \rightarrow \infty} \left( \frac{x+5}{x-7} \right)^x$  is
- (a)  $e^{12}$   
 (b)  $e^{-12}$   
 (c) 12  
 (d) -12

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**ANSWER KEY**

|    |   |    |   |    |   |
|----|---|----|---|----|---|
| 1  | c | 21 | c | 41 | d |
| 2  | a | 22 | c | 42 | d |
| 3  | c | 23 | b | 43 | c |
| 4  | b | 24 | a | 44 | b |
| 5  | b | 25 | a | 45 | d |
| 6  | a | 26 | c | 46 | d |
| 7  | c | 27 | a | 47 | b |
| 8  | c | 28 | d | 48 | b |
| 9  | c | 29 | c | 49 | a |
| 10 | a | 30 | c | 50 | a |
| 11 | c | 11 | a | 51 | d |
| 12 | d | 12 | b | 52 | a |
| 13 | d | 13 | c | 53 | c |
| 14 | b | 14 | b | 54 | c |
| 15 | a | 15 | a | 55 | c |
| 16 | a | 16 | d | 56 | d |
| 17 | b | 17 | a | 57 | b |
| 18 | c | 18 | c | 58 | c |
| 19 | c | 19 | c | 59 | c |
| 20 | b | 20 | c | 60 | a |

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